Liquid-liquid phase transitions and water-like anomalies

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Departmental seminar

with

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Content

- Introduction to liquid-liquid phase transition (LLPT) and anomalies
- Spherical water model
- ST2 model
- Locating the liquid-liquid critical point (LLCP)
- Conclusions

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Water!

- Water is most common liquid on Earth
- Many experiments done on water
- Water has many unusual properties (69 anomalies)

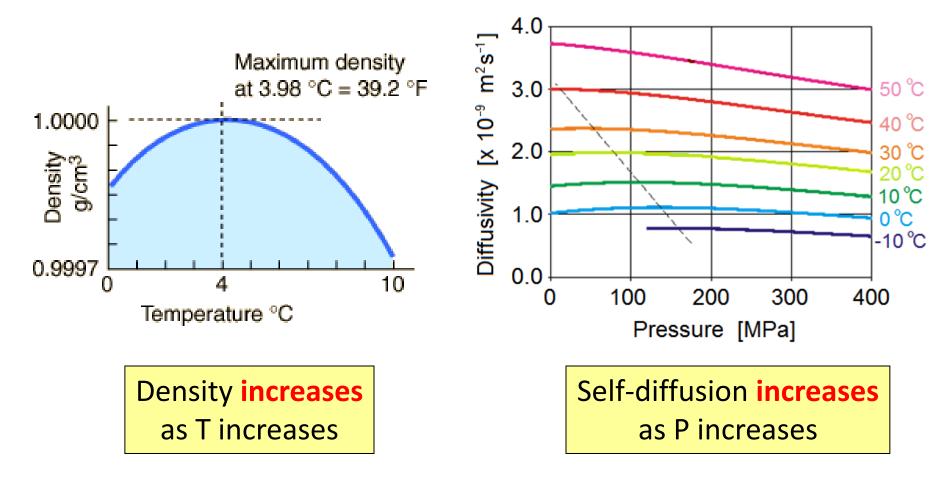
www.lsbu.ac.uk/water/anmlies.html



Water anomalies

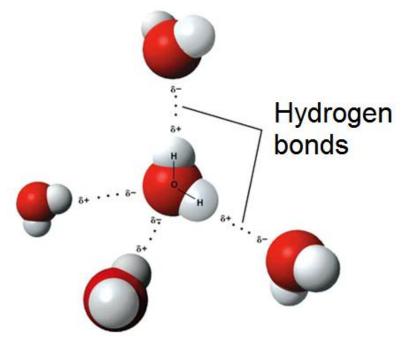
Density anomaly

Diffusion anomaly



Water anomalies explained?

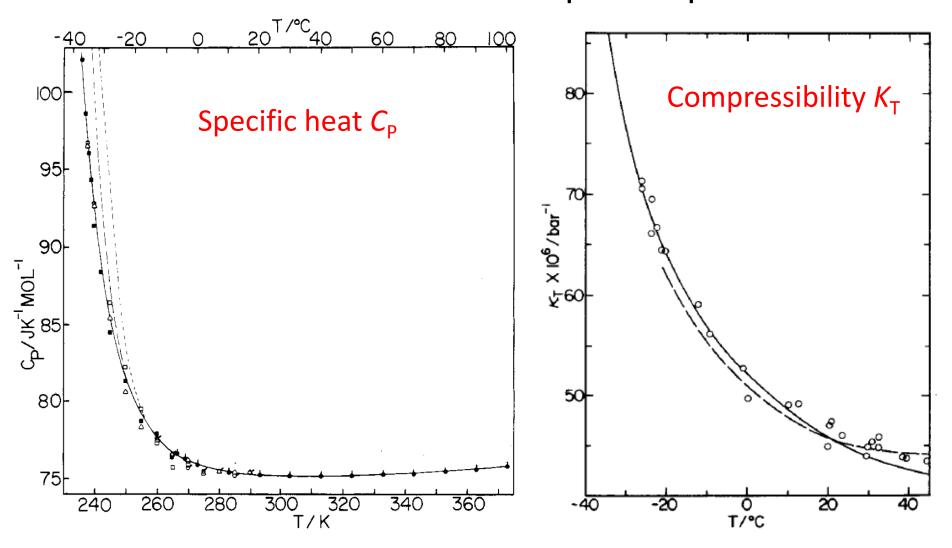
 Are anomalies caused solely by H-bonds?



• No!

Other liquids without H-bonds also show same anomalies!

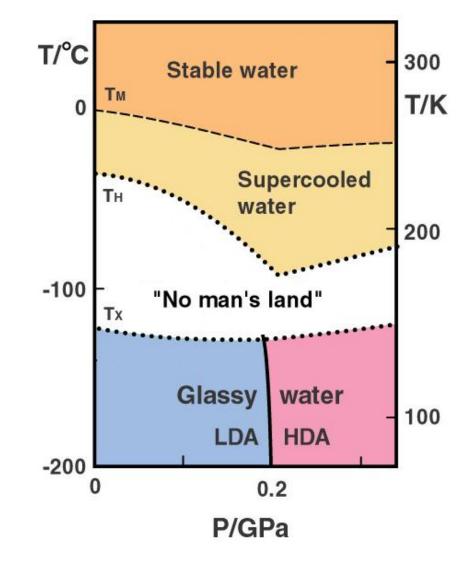
Supercooled water: C_{P} & K_{T} diverge



Amorphous ice: HDA & LDA





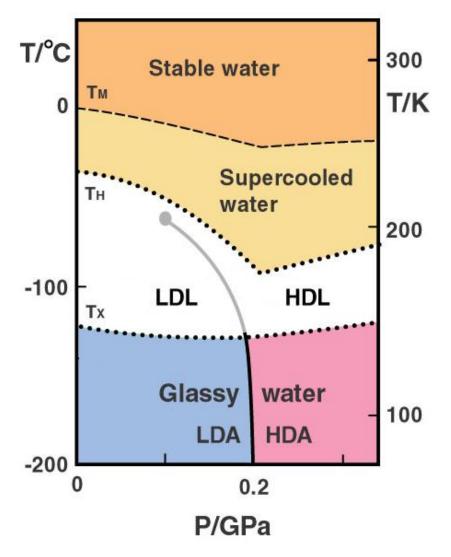


Experiments by Katrin Winkel, et al.

LLPT hypothesis for water

(Poole/Sciortino/Essmann/Stanley, 1992)

- Two liquids below nucleation temperature:
 - Low density liquid (LDL)
 - High density liquid (HDL)
- Separated by liquid-liquid phase transition
- Ending in liquid-liquid critical point



Main question

 Main question: where do anomalies come from and how are they related to LLPT / LLCP?

• Start with a simple model!

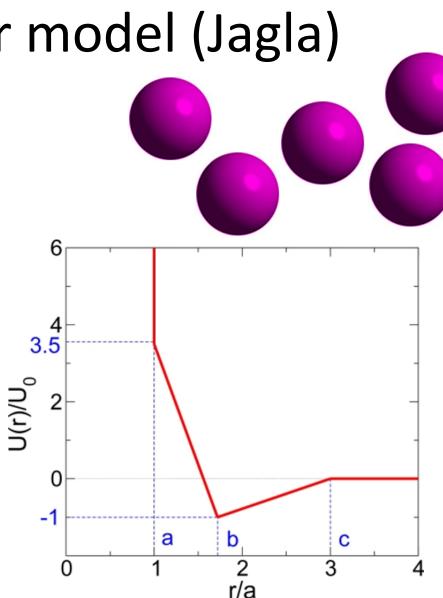
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Spherical water model (Jagla)

Monatomic particles

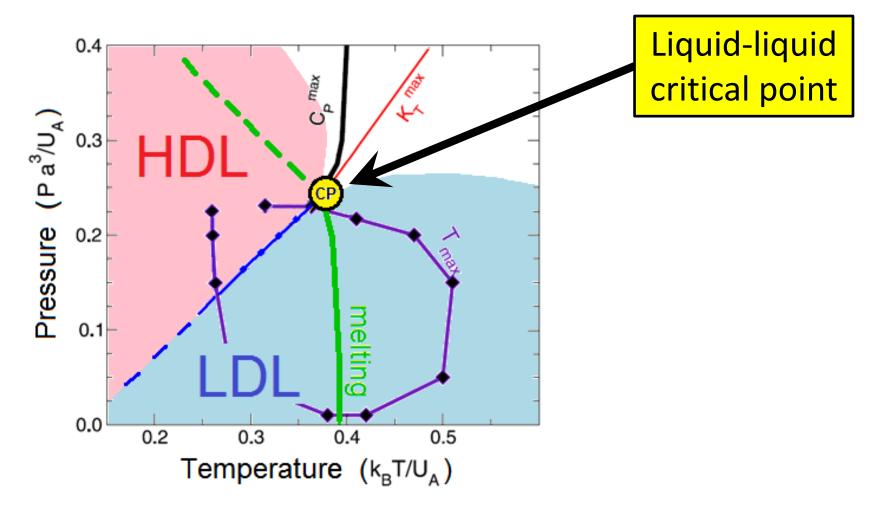
• Pairwise interaction

Hard core
+ soft core
+ potential well



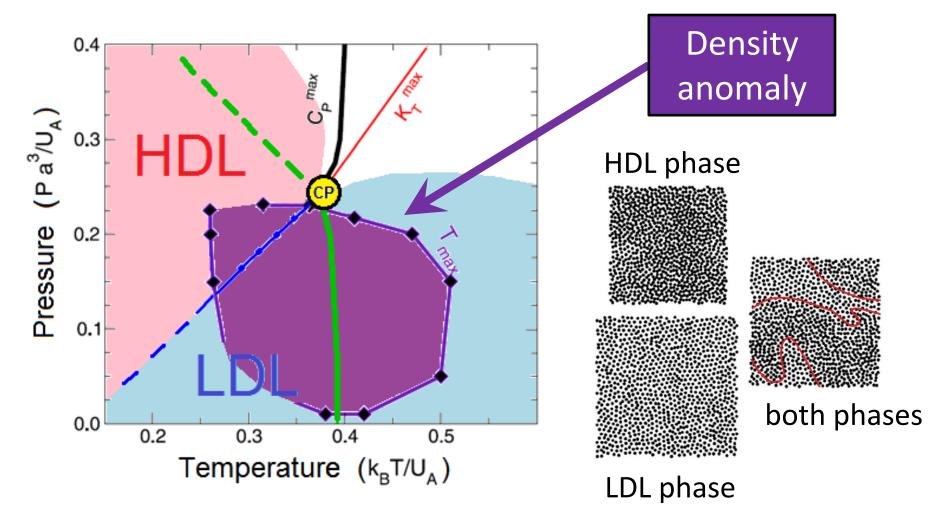
E. A. Jagla, J. Chem. Phys. **111**, 8980 (1999)

Phase diagram: LLPT & anomalies



L. Xu, S. Buldyrev, C. A. Angell, H. E. Stanley, Phys. Rev. E 74, 031108 (2006)

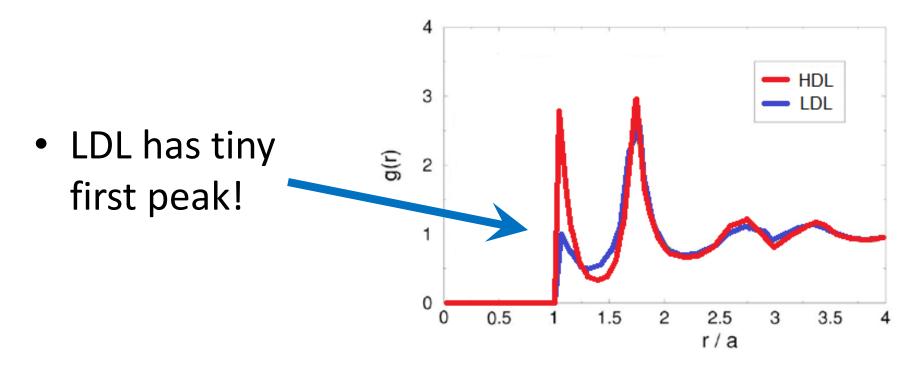
Phase diagram: LLPT & anomalies



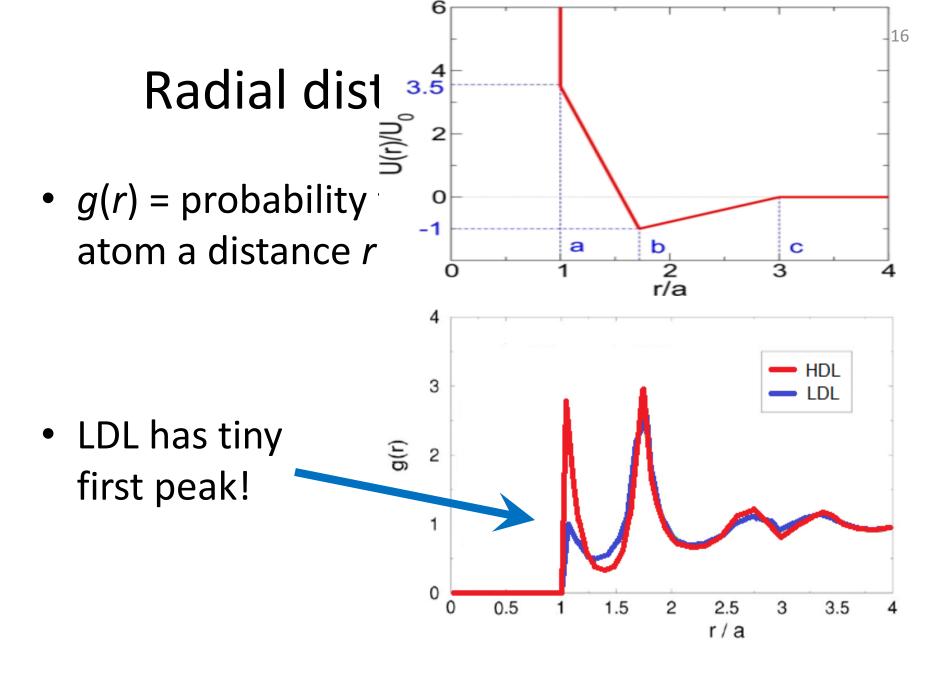
L. Xu, S. Buldyrev, C. A. Angell, H. E. Stanley, Phys. Rev. E 74, 031108 (2006)

Radial distribution function

 g(r) = probability for atom to find another atom a distance r away



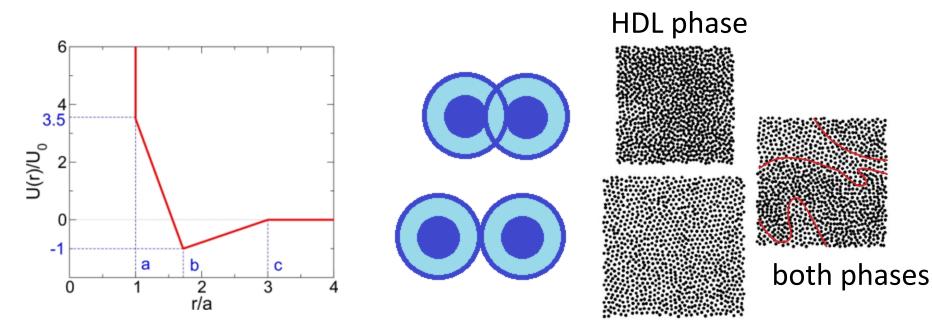
L. Xu, S. Buldyrev, C. A. Angell, H. E. Stanley, Phys. Rev. E 74, 031108 (2006)



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Two competing length scales

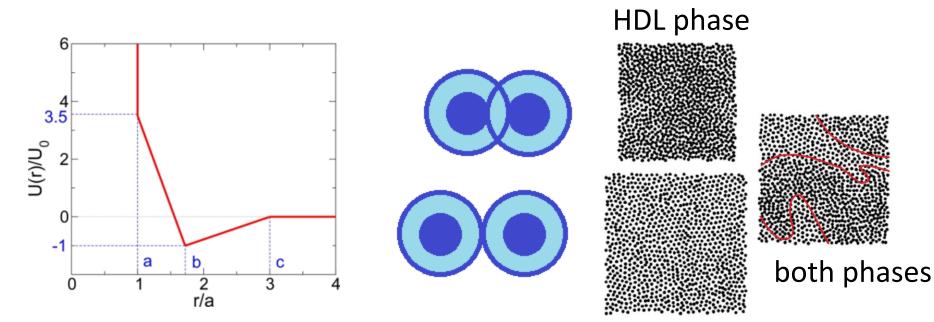
• Two liquids (LDL and HDL)



LDL phase

Two competing length scales

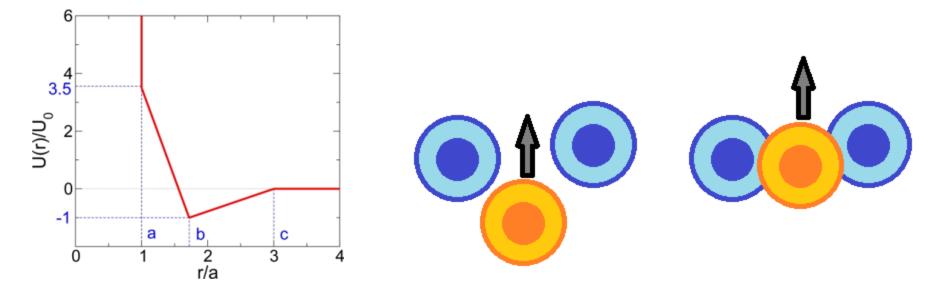
- Two liquids (LDL and HDL)
- Higher T leads to overlap → density anomaly



LDL phase

Two competing length scales

- Two liquids (LDL and HDL)
- Higher T leads to overlap \rightarrow density anomaly
- Higher P leads to overlap → diffusion anomaly



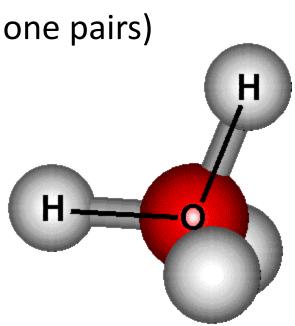
Conclusion spherical water model

- Origin of anomalies and LLPT are clear in this model: two competing length scales
- Now consider more realistic models!

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ST2 water model

- Each water molecule has 5 "atoms"
- 4 charges (2 hydrogen atoms & 2 lone pairs) with electrostatic interaction
- 1 oxygen atom with Lennard-Jones interaction

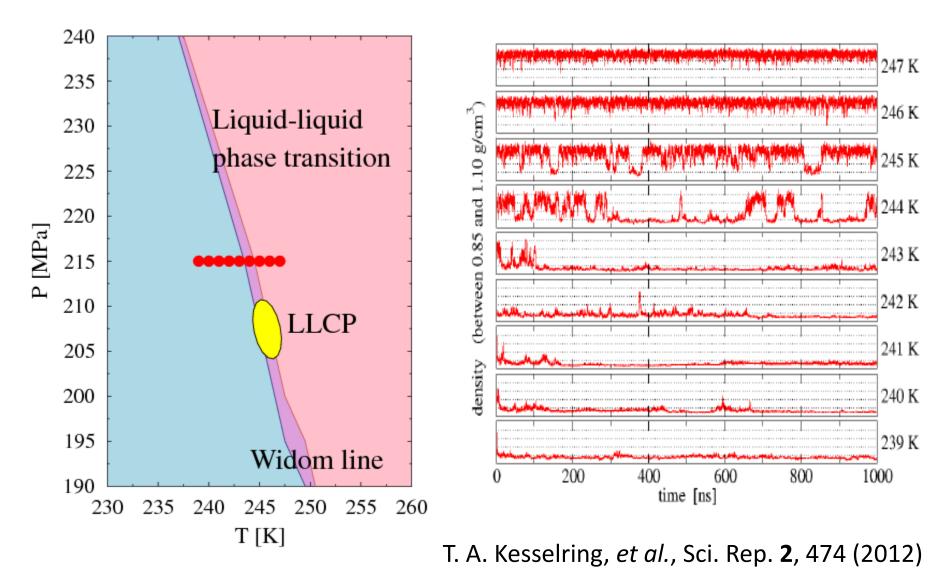


ST2 phase diagram 300 HDL T of density max 200 LLCP K_T max LDL 100 C_P max P (MPa) Widom line 0 -100 -200 275 250 225 300 325 350 T (K)

Poole, et al., JPCM 17, L431 (2005)

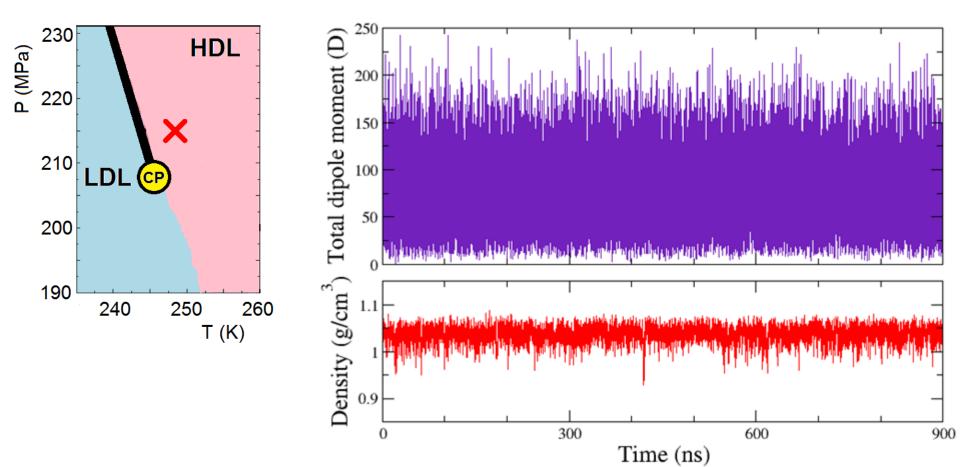
"Phase flipping" in NPT ensemble

24



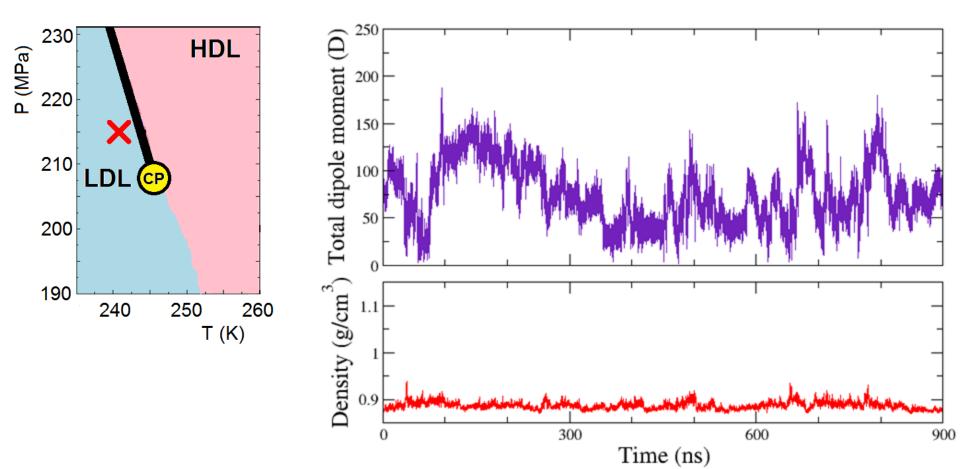
Total dipole moment in HDL

343 molecules / 215 MPa / 247 K

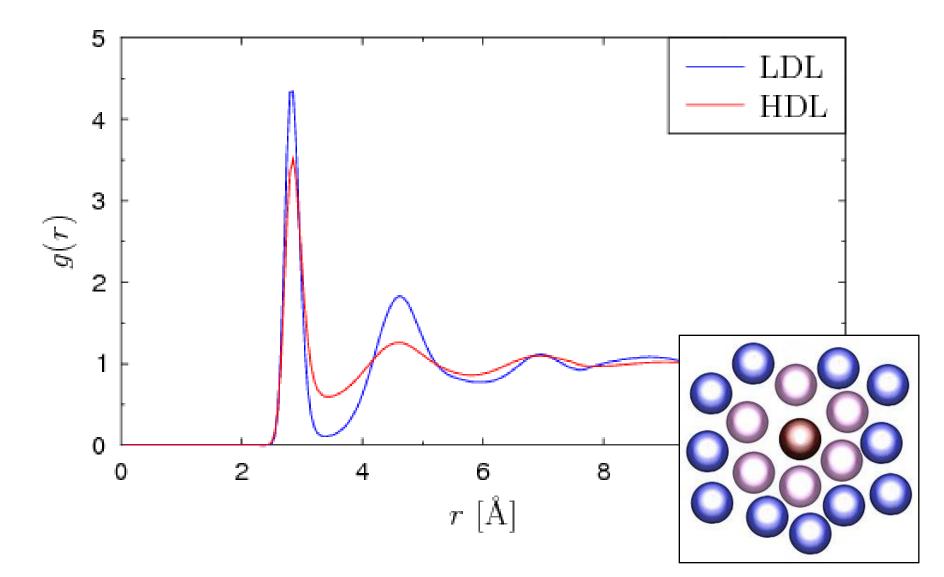


Total dipole moment in LDL

343 molecules / 215 MPa / 241 K



ST2 radial distribution function

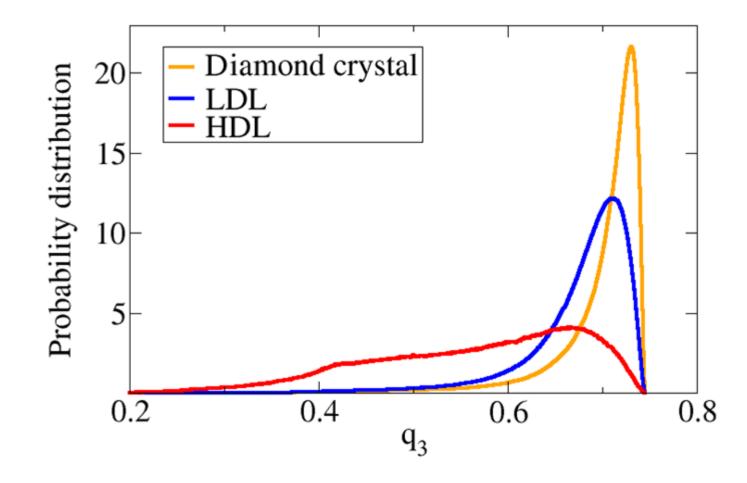


Structural parameter q_3 $q_{\ell}(i) \equiv \sum_{m} \frac{1}{N} \sum_{j \in n(i)} Y_{\ell}^m(\varphi_{ij}, \vartheta_{ij}) \quad -\ell \leq m \leq \ell$

- 1. Calculate spherical harmonics Y_{ℓ}^{m} for each pair (i,j)
- 2. Average over all neighbors n(i) of molecule i
- 3. Sum over all *m*
- Each molecule has its own number
- For perfect diamond: $q_3 \approx 0.73$
- For completely random: $q_3 = 0$

$$Y_0^0(\theta,\varphi) = \frac{1}{2}\sqrt{\frac{1}{\pi}}$$
$$Y_1^{-1}(\theta,\varphi) = \frac{1}{2}\sqrt{\frac{3}{2\pi}}\sin\theta \, e^{-i\varphi}$$
$$Y_1^0(\theta,\varphi) = \frac{1}{2}\sqrt{\frac{3}{\pi}}\cos\theta$$
$$Y_1^1(\theta,\varphi) = \frac{-1}{2}\sqrt{\frac{3}{2\pi}}\sin\theta \, e^{i\varphi}$$

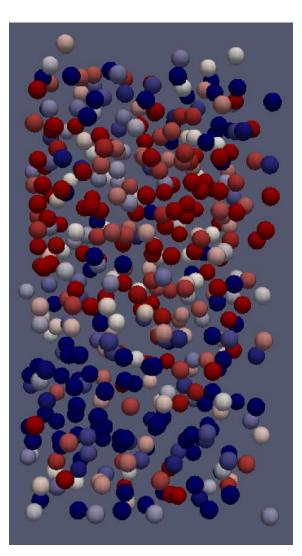
q_3 histogram



Again: LDL and HDL have clearly different structures

Phase segregation

- Structural parameters such as q₃ allow for real-time determination of local structure!
- Phase segregation can be witnessed in simulations



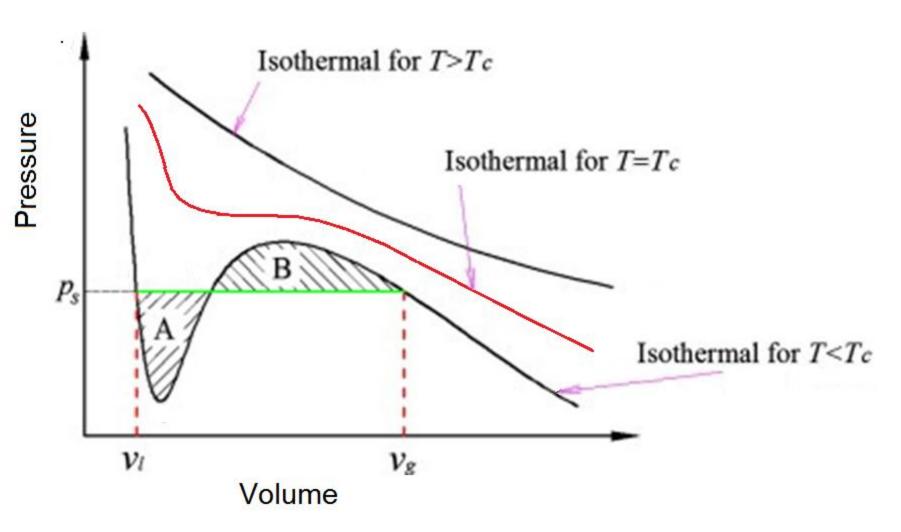
Conclusion ST2

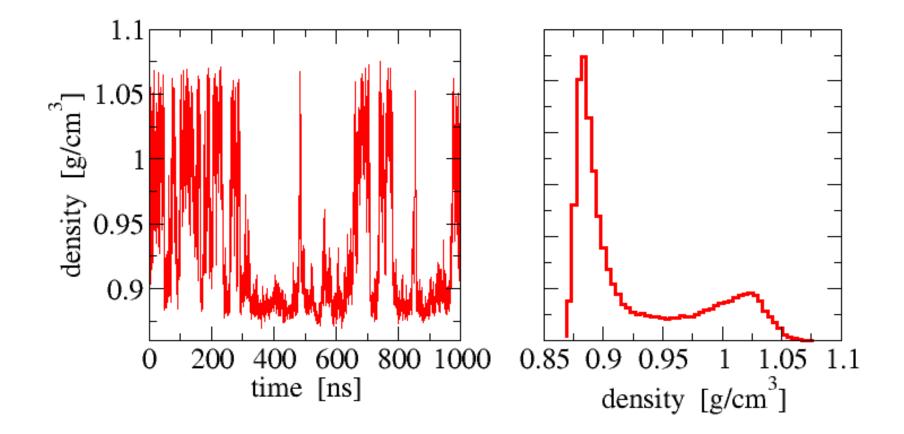
• ST2 allows for two liquid structures to exist simultaneously (at same P, T)

• This leads to LLPT and anomalies

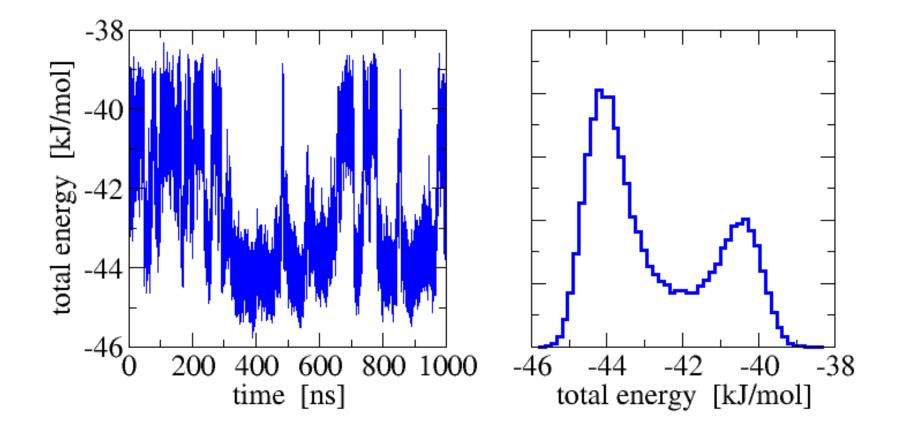
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Via PV diagram

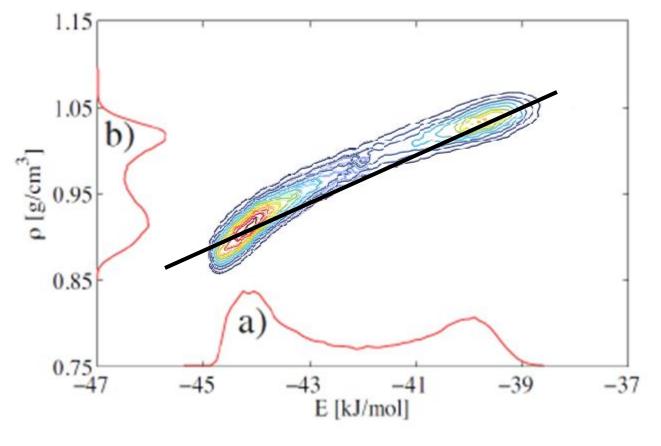




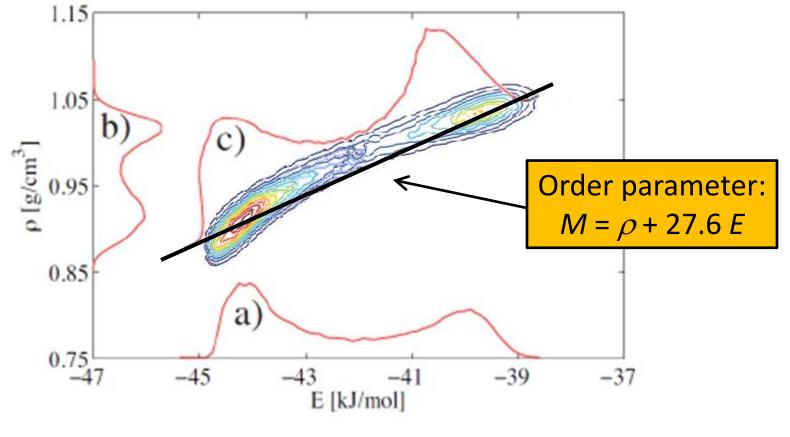
Phase flipping \rightarrow histogram of density



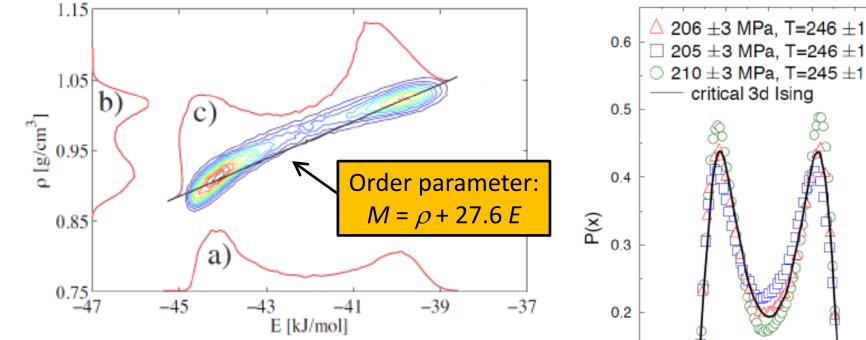
Phase flipping \rightarrow histogram of energy



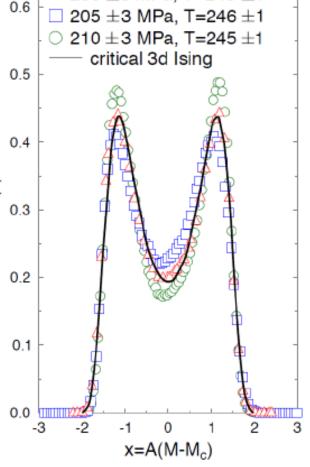
2D histogram of energy & density



2D histogram of energy & density



Order parameter *M* can be fitted to 3D Ising model



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Conclusions

- Simple models can have LLPT + anomalies because of two length scales and therefore two liquid structures
- Atomic models can have LLPT + anomalies because of two liquid structures
- LLCP falls in 3D Ising universality class which can be used to locate LLCP accurately

